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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/664,446	09/18/2003	Takatomo Nishino	09792909-5672	4529
26263	7590	08/22/2007	EXAMINER	
SONNENSCHEIN NATH & ROSENTHAL LLP			DOVE, TRACY MAE	
P.O. BOX 061080			ART UNIT	PAPER NUMBER
WACKER DRIVE STATION, SEARS TOWER			1745	
CHICAGO, IL 60606-1080				
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08/22/2007		PAPER		

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Office Action Summary	Application No.	Applicant(s)	
	10/664,446	NISHINO ET AL.	
	Examiner	Art Unit	
	Tracy Dove	1745	

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --
Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) OR THIRTY (30) DAYS, WHICHEVER IS LONGER, FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

- 1) Responsive to communication(s) filed on 05 July 2007.
- 2a) This action is **FINAL**. 2b) This action is non-final.
- 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

- 4) Claim(s) 1-12 is/are pending in the application.
 - 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
- 5) Claim(s) _____ is/are allowed.
- 6) Claim(s) 1-12 is/are rejected.
- 7) Claim(s) _____ is/are objected to.
- 8) Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

- 9) The specification is objected to by the Examiner.
- 10) The drawing(s) filed on _____ is/are: a) accepted or b) objected to by the Examiner.
 Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
 Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
- 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

- 12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
 - a) All b) Some * c) None of:
 1. Certified copies of the priority documents have been received.
 2. Certified copies of the priority documents have been received in Application No. _____.
 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

- | | |
|--|---|
| 1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892) | 4) <input type="checkbox"/> Interview Summary (PTO-413) |
| 2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948) | Paper No(s)/Mail Date. _____ . |
| 3) <input type="checkbox"/> Information Disclosure Statement(s) (PTO/SB/08) | 5) <input type="checkbox"/> Notice of Informal Patent Application |
| Paper No(s)/Mail Date _____ . | 6) <input type="checkbox"/> Other: _____ . |

DETAILED ACTION

This Office Action is in response to the communication filed on 7/5/07. Applicants arguments have been considered, but are not persuasive. Claims 1-12 are pending.

Continued Examination Under 37 CFR 1.114

A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 7/5/07 has been entered.

Priority

Acknowledgment is made of applicant's claim for priority under 35 U.S.C. 119(a)-(d) based upon an application filed in Japan on 9/17/02. A claim for priority under 35 U.S.C. 119(a)-(d) cannot be based on said application, since the United States application was filed more than twelve months thereafter. Examiner points out that while the preliminary amendment filed on 9/18/03 has amended the specification to recite "priority under 35 U.S.C. 119 is not claimed", the declaration/oath filed on 3/8/04 does not indicate priority is not being claimed (box on page 2 is not checked). Appropriate correction is required.

Receipt is acknowledged of papers submitted under 35 U.S.C. 119(a)-(d), which papers have been placed of record in the file.

Claims Analysis

Claim 1 recites "effected by applying a compressive force and a shearing force to at least a part of a surface of a base material when the composite material is formed", which is not given

patentable weight because it is a product-by-process limitation. Claim 7 also recites this limitation.

Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C. 102 that form the basis for the rejections under this section made in this Office action:

A person shall be entitled to a patent unless –

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

(e) the invention was described in a patent granted on an application for patent by another filed in the United States before the invention thereof by the applicant for patent, or on an international application by another who has fulfilled the requirements of paragraphs (1), (2), and (4) of section 371(c) of this title before the invention thereof by the applicant for patent.

The changes made to 35 U.S.C. 102(e) by the American Inventors Protection Act of 1999 (AIPA) and the Intellectual Property and High Technology Technical Amendments Act of 2002 do not apply when the reference is a U.S. patent resulting directly or indirectly from an international application filed before November 29, 2000. Therefore, the prior art date of the reference is determined under 35 U.S.C. 102(e) prior to the amendment by the AIPA (pre-AIPA 35 U.S.C. 102(e)).

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

Claims 1-12 are rejected under 35 U.S.C. 102(b)/103(a) as being anticipated by, and alternatively unpatentable over, Kawakami et al., US 6,432,585.

Kawakami teaches a battery comprising an anode, a cathode and an electrolyte. The anode comprises an anode structural body 10. The structural body comprises a host material 101

in an amount of 50 wt% or more. If the electrode structural body is used in a lithium battery, the host material comprises one or more elements selected from the group consisting of Si, Sn and In (11:1-18). When Si is used as the host material, Cu, Ni, Ag or Sn may partially cover the surface of the Si particles (11:30-67). Si may contain an impurity such as Al, Ca, Cr, Fe, Mg, Mn or Ni to decrease the electric resistance of the electrode material layer 102 (12:1-5). The layer 102 may comprise the host material 101 and an electrically conductive auxiliary in order to assist and increase the electron conduction among particles of the host material or that between the host material and the collector. It is preferred the electrically conductive auxiliary be contained in an amount of 1-30 wt%. The electrically conductive auxiliary may be a carbonaceous material such as acetylene black, ketjen black or graphite. The electrically conductive auxiliary may be in a filament-like, fibrous or needle-like form. The host material and carbonaceous material are mechanically mixed using a ball mill or the like (compressive/shearing force) (12:46-13:9). See also column 19, line 50-column 20, line 23. See also Example 12.

Thus the claims are anticipated. Kawakami does not explicitly state the host material and carbonaceous material are bonded by van der Waals forces. However, when the host material and carbon material are mechanically mixed using a ball mill or the like (compressive/shearing force), the host material and carbon material are inherently attracted by van der Waals forces (weak attractive forces acting between molecules; see Hawley's Condensed Chemical Dictionary, page 1217). In order for the carbon material to assist and increase the electron conduction among particles of the host material, the carbon material must be in contact with the host material.

*

Claims 1, 3-7 and 9-12 are rejected under 35 U.S.C. 102(b)/103(a) as being anticipated by, and alternatively unpatentable over, Suzuki et al., US 6,413,672.

Suzuki teaches a lithium battery comprising an anode, a cathode and an electrolyte. The anode comprises an anode material containing 50-99 wt% of silicon and 1-50 wt% of carbon material (abstract). It is preferable that silicon exists in the form of particles and the particles are covered with the carbonaceous material. A material containing a high proportion of silicon provides a high capacity (2:52-67). An amount of 70 wt% or more of silicon is preferred (5:7-9). The carbonaceous material may be graphite, amorphous carbon (acetylene black) or a mixture thereof. For example, coke, natural graphite, artificial graphite, carbonized pitch or a mixture thereof may be used (5:16-22). Embodiment 1 teaches 28.5 parts by weight silicon and 7 parts by weight graphite were mixed and then processed in a vibration mill (compressive/shearing).

Thus the claims are anticipated. The carbon covering material is inherently bonded to the silicon by van der Waals forces (weak attractive forces acting between molecules; see Hawley's Condensed Chemical Dictionary, page 1217) when the material is subjected to a vibration mill.

*

Claims 1, 4, 5, 7, 10 and 11 are rejected under 35 U.S.C. 102(b)/103(a) as being anticipated by, and alternatively unpatentable over, Suzuki et al., US 6,171,725.

Suzuki teaches a battery comprising a positive electrode, a negative electrode and an electrolyte. The negative electrode includes a negative electrode material containing 30-90 wt% of silicon and 10-70 wt% of carbon (abstract). The carbon material may be cokes, graphite (artificial graphite) and the like (3:14-21). The silicon/carbon composite material preferably comprises 50-90 wt% silicon and 10-50 wt% carbon (3:22-63). Example 4 teaches silicon

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powder was mixed with graphite/pitch. After calcining, the solid material was roughly milled (compressed/sheared). Through dry milling, a silicon/carbon composite powder was obtained.

Thus the claims are anticipated. The teaching of a silicon/carbon composite material obtained by dry milling inherently teaches the limitation “physically bonded by van der Waals forces”. The teaching of a composite material clearly indicates the silicon and carbon materials are bonded.

*

Claims 1-12 are rejected under 35 U.S.C. 102(e)/103(a) as being anticipated by Inoue et al., US 6,506,520.

Inoue teaches a negative electrode for a nonaqueous secondary battery comprising composite particles (abstract). The composite particles include a core phase A and an outer phase B. When phase A is Sn, phase B may be Sn-Fe, Sn-Zn, Sn-In or Sn-Pb. When phase A is Si, phase B may be Si-Co, Si-Ni, Si-Zn or Si-Al (Table 1). A conductive material may be contained in the negative electrode. Among conductive materials, synthetic (artificial) graphite, acetylene black and carbon fibers are especially favorable. The amount of conductive material in the negative electrode is preferably 1-30% of the negative electrode materials (composite particles) (5:50-6:3).

Thus the claims are anticipated. The claims are alternatively unpatentable because the courts have ruled that product-by-process limitations, in the absence of unexpected results are obvious. Inoue does not explicitly state a compressive and/or shearing force is applied to the negative electrode material, however, the negative electrode material of the claimed invention and the negative electrode of the prior art appear to be the same.

Response to Arguments

Applicant's arguments filed 7/5/07 have been fully considered but they are not persuasive.

Applicant asserts, unlike the prior art, the claimed use of the compressive force and shearing action to combine the base material and the Group 14 element produces a composite material that is physically bonded by van der Waals forces. However, the claimed anode material must be shown to be materially different than the anode material of the prior art. Applicant has not provided any evidence clearly showing the claimed material is materially different than the prior art material. Applicant's asserted unexpected results do not properly compare the claimed invention with the prior art of record.

Applicant states none of the cited references discloses or fairly suggest particles or a material resulting from the application of a compressive force and shearing action as claimed. However, this limitation is a product-by-process limitation that is obvious in the absence of unexpected results. The claimed material and the material of the prior art appear to be the same.

Furthermore, the mechanical mixing using a ball mill or the like of Kawakami results in compressive and shearing forces. The mixing in a vibration mill or roughly milling of Suzuki '672 or Suzuki '725 results in compressive and shearing forces. Examiner points out Figure 1 of the present specification as an example of the claimed compressive/shearing forces. It is unclear how Figure 1, which uses a compressing bar is different than, for example, the ball mill of the prior art that uses compressing balls. It is unclear how Applicant concludes the compressing bar of Figure 1 results in compressive/shearing forces and the compressing balls of the prior art do not result in compressive/shearing forces.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Tracy Dove whose telephone number is 571-272-1285. The examiner can normally be reached on Monday-Thursday (9:00-7:30).

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Pat Ryan can be reached on 571-272-1292. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

August 20, 2007



TRACY DOVE
PRIMARY EXAMINER